

Name _____

Ions and Formulas Test Practice

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| 1. Chromic | 31. Phosphate | 61. Oxide |
| 2. Fluoride | 32. Mercuric | 62. He |
| 3. Sodium | 33. Sodium Hydroxide | 63. Glucose |
| 4. Mercury (II) | 34. Neon | 64. Cupric |
| 5. Calcium | 35. Water | 65. Nitrate |
| 6. Chloride | 36. Chromium (III) | 66. Silver |
| 7. Ammonia | 37. Barium | 67. Sulfite |
| 8. Iodide | 38. Nitric Acid | 68. Carbon Dioxide |
| 9. Sulfide | 39. Chlorite | 69. Ferric |
| 10. Hydrogen Sulfate | 40. Carbon Monoxide | 70. Carbonate |
| 11. Chlorate | 41. Magnesium | 71. Potassium |
| 12. Copper (II) | 42. Chromate | 72. Carbon tetrachloride |
| 13. Hydrogen | 43. Dichromate | 73. Mercury (I) |
| 14. Peroxide | 44. Aluminum | 74. Hypochlorite |
| 15. Hydroxide | 45. Iron (III) | 75. Lead (IV) |
| 16. Hydrogen Carbonate | 46. Hydrogen ion | 76. Sulfate |
| 17. Ferrous | 47. Methane | 77. Permanganate |
| 18. Oxygen | 48. Cuprous | 78. Nickelic |
| 19. Tin (IV) | 49. Ammonium | 79. Lead (II) |
| 20. Perchlorate | 50. Stannous | 80. Proton |
| 21. Beryllium | 51. Nitrite | 81. Sucrose |
| 22. Zinc | 52. Hydronium | 82. Chromium (II) |
| 23. Copper (I) | 53. Nitrogen | 83. Sulfuric Acid |
| 24. Iron (II) | 54. Bromine | 84. Hydrochloric Acid |
| 25. Plumbous | 55. Nickelous | 85. Fluorine |
| 26. Stannic | 56. Mercurous | 86. Chlorine |
| 27. Sodium Chloride | 57. Plumbic | 87. Iodine |
| 28. Nickel (III) | 58. Acetate | 88. Lithium |
| 29. Nickel (II) | 59. Bromide | 89. Tin (II) |
| 30. Chromous | 60. Diarsenic pentoxide | |